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NEWS	5	AUG	20	CA/CAplus enhanced with CAS indexing in pre-1907 records
NEWS	6	AUG	27	Full-text patent databases enhanced with predefined patent family display formats from INPADOCDB
NEWS	7	AUG	27	USPATOLD now available on STN
NEWS	8	AUG		CAS REGISTRY enhanced with additional experimental
		*****		spectral property data
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NEWS	1.0	SEP	13	FORIS renamed to SOFIS
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NEWS	15	OCT	02	CA/CAplus enhanced with pre-1907 records from Chemisches Zentralblatt
NEWS	16	OCT	19	BEILSTEIN updated with new compounds
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NEWS		DEC		TOXCENTER enhanced with 2008 MeSH vocabulary in
				MEDLINE segment
NEWS	26	DEC	17	MEDLINE and LMEDLINE updated with 2008 MeSH vocabulary
NEWS	27	DEC	17	CA/CAplus enhanced with new custom IPC display formats
NEWS	28	DEC	17	STN Viewer enhanced with full-text patent content from USPATOLD
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NEWS	32	JAN	28	MARPAT searching enhanced
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NEWS 35 JAN 28 MEDLINE and LMEDLINE reloaded with enhancements

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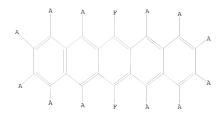
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L1 STRUCTURE UPLOADED

=> d 11 L1 HAS NO ANSWERS

L1 STR



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=> s 11

SAMPLE SEARCH INITIATED 09:29:59 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 81 TO ITERATE

100.0% PROCESSED 81 ITERATIONS

0 ANSWERS

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BATCH \*\*COMPLETE\*\*

PROJECTED ITERATIONS: 1081 TO 2159
PROJECTED ANSWERS: 0 TO 0

L2 0 SEA SSS SAM L1

=> s 11 full

FULL SEARCH INITIATED 09:30:03 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1577 TO ITERATE

100.0% PROCESSED 1577 ITERATIONS SEARCH TIME: 00.00.01 11 ANSWERS

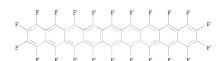
L3 11 SEA SSS FUL L1

=> d 13 scan

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

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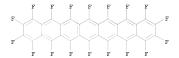
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HOW MANY MORE ANSWERS DO YOU WISH TO SCAN? (1):5

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN INDEX NAME NOT YET ASSIGNED

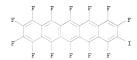
MF C30 F18



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- L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN
- IN Pentacene, 1,2,3,4,5,6,7,8,9,11,12,13,14-tridecafluoro-10-iodo-

MF C22 F13 I



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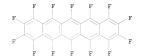
- L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN
- IN Pentacene, 2-chloro-1,3,4,5,6,7,8,9,10,11,12,13,14-tridecafluoro-

MF C22 C1 F13

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L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro-MF C22 F14



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- T. 3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN
- IN INDEX NAME NOT YET ASSIGNED
- MF C42 F24



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L6 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN 2004:577543 CAPLUS

ACCESSION NUMBER:

DOCUMENT NUMBER: 141:260225

TITLE: The essential role of H-F substitution in the

electron-phonon interactions and electron transfer in

the negatively charged acenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto University, Sakyo-ku, Kyoto, 606-8103, Japan

Journal of Chemical Physics (2004), 121(5), 2356-2366

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE:

SOURCE:

English The single charge transfer through acenes, partially H-F substituted acenes, and fluoroacenes is discussed. The reorganization energies between the neutral mols. and the corresponding monoanions for partially H-F substituted acenes lie between those for acenes and fluoroacenes. The delocalization of the lowest unoccupied MOs (LUMO) by substituting hydrogen atoms by fluorine atoms with the highest electronegativity in every element is the main reason why the reorganization energy between the neutral mol. and the monoanion for partially H-F substituted acenes lies between those for acenes and fluoroacenes. This result implies that the neq. charged partially H-F substituted acenes would be better conductors with rapid electron transfer than the neg. charged fluoroacenes if we assume that the overlap of the LUMO between partially H-F substituted acenes is not significantly different from that between two neighboring fluoroacenes. The structures of the monoanions of acenes, fluoroacenes, and partially H-F substituted acenes are optimized under D2h geometry, and the Jahn-Teller effects in the monoanions of benzene and fluorobenzene are discussed. The vibration effect onto the charge transfer problem is also discussed. The C-C stretching modes around  $1500~\mathrm{cm}{-}1$  are the main modes converting the neutral mols. to the monoanions in acenes, fluoroacenes, and partially H-F substituted acenes. It can be confirmed from the calculational results that the C-C stretching modes around 1500 cm-1 the most strongly couple to the LUMO in these mols. The main reason why the total electron-phonon coupling consts. (ILUMO) for the monoanions of acenes in which four outer hydrogen atoms are substituted by fluorine atoms are larger than those for the monoanions of acenes in which several inner hydrogen atoms are substituted by fluorine atoms is suggested. The

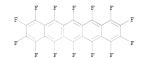
relationships between the electron transfer and the electron-phonon interactions are discussed. The plot of the reorganization energies against the LLUMO values is found to be nearly linear. In view of these results, the relationships between the normal and superconducting states are briefly discussed.

646533-88-2 690975-12-3 RL: PRP (Properties)

(essential role of H-F substitution in electron-phonon interactions and electron transfer in neg. charged acenes)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)

REFERENCE COUNT: 99 THERE ARE 99 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:480070 CAPLUS

DOCUMENT NUMBER: 141:197895

TITLE: Perfluoropentacene: High-Performance p-n Junctions and

Complementary Circuits with Pentacene
AUTHOR(S): Sakamoto, Youichi; Suzuki, Toshiyasu; Kobayashi,

Masafumi; Gao, Yuan; Fukai, Yasushi; Inoue, Youji; Sato, Fumio; Tokito, Shizuo

Institute for Molecular Science, Okazaki, 444-8787,

Japan

Journal of the American Chemical Society (2004),

126(26), 8138-8140 CODEN: JACSAT: ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

CORPORATE SOURCE:

SOURCE:

LANGUAGE: English

AB The authors report the synthesis and characterization of

perfluoropentacene as an n-type semiconductor for organic field-effect transistors (OFETs). Perfluoropentacene is a planar and crystalline material that adopts a herringbone structure as observed for pentacene. OFETs with perfluoropentacene were constructed using top-contact geometry, and an electron mobility of 0.11 cm2 V-1 s-1 was observed Bipolar OFETs with

perfluoropentacene and pentacene function at both neg. and pos. gate voltages. The improved p-n junctions are probably due to the similar d-spacings of both acenes. Complementary inverter circuits were fabricated, and the transfer characteristics exhibit a sharp inversion of the output signal with a high-voltage gain.

T 646533-88-2P, Perfluoropentacene RL: DEV (Device component use); PNU (Preparation, unclassified); PREP

(perfluoropentacene: high-performance p-n junctions and complementary circuits with pentacene)

RN 646533-88-2 CAPLUS

(Preparation); USES (Uses)

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



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L6 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:297458 CAPLUS

DOCUMENT NUMBER: 140:429323

TITLE: Inverse isotope effects and electron-phonon coupling in the positively charged deutero- and fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto

University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 120(16), 7659-7672 CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal

LANGUAGE: English

Electron-phonon interactions in the monocations of deutero- and fluoroacenes are studied and compared with those in the monocations of acenes and those in the monoanions of fluoroacenes. Because of the significant phase pattern difference between the highest occupied MOs (HOMO) and the lowest unoccupied MOs (LUMO), the frequency modes lower than 500 cm-1 and the high-frequency modes around 1400 cm-1 couple more strongly to the LUMO than to the HOMO, while the frequency modes around 500 cm-1 and the frequency modes around 1600 cm-1 couple more strongly to the HOMO than to the LUMO in fluoroacenes with D2h geometry. The total electron-phonon coupling consts. for the monocations (1HOMO) are estimated and compared with those for the monoanions (1LUMO) in deutero- and fluoroacenes. The 1HOMO values are estimated to be 0.418, 0.399, 0.301, 0.255, and 0.222 eV for C6F6 (if), CloF8 (2f), C14F10 (3f), C18F12 (4f), and C2F14 (5f), resp. The HOMO values are smaller than the LLDMO value of smaller than the LDMO values are smaller than the LDMO values. mol. size less rapidly than the 1LUMO value in fluoroacenes, and the 1HOMO value of 0.074 eV is much larger than the 1LUMO value of 0.009 eV in polyfluoroacene. The logarithmically averaged phonon frequencies for the monocations (@ln,HOMO) are estimated to be larger than those for the monoanions (@ln,LUMO) in fluoroacenes. This is because the C-C stretching modes around 1600 cm-1 couple most strongly to the HOMO, and those around 1400 cm-1 couple the most strongly to the LUMO in

fluoroacenes. The significant phase pattern difference between the HOMO and the LUMO is the main reason for the calculational results. The 1HOMO values increase much more significantly by H-F substitution than by H-D substitution in acenes. The possible inverse isotope effects in the electron-phonon interactions as a consequence of deuteration in the monocations of nanosized mols, are suggested.

690975-12-3

RL: PRP (Properties) (inverse isotope effects and electron-phonon coupling in pos. charged deutero- and fluoroacenes)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT:

SOURCE:

PUBLISHER:

38 THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 4 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:898555 CAPLUS

DOCUMENT NUMBER: 140:118999

TITLE:

Electron-phonon interactions in the monoanions of fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Graduate School of Engineering, Department of Material

Chemistry, Kyoto University, Sakvo-ku, Kyoto,

606-8501, Japan Journal of Chemical Physics (2003), 119(21),

11318-11328

CODEN: JCPSA6; ISSN: 0021-9606 American Institute of Physics

DOCUMENT TYPE: Journal

LANGUAGE: English AB

Electron-phonon interactions in the monoanions of fluoroacenes such as C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f) are studied, and compared with those in the monoanions of acenes and deutero-acenes. The C-C stretching modes around 1500 cm-1 the most strongly couple to the lowest unoccupied MOs (LUMO) in fluoroacenes. The estimated total electron-phonon coupling consts. (ILUMO) are 0.475, 0.473, 0.350, 0.273, and 0.215 eV for 1f, 2f, 3f, 4f, and 5f, resp. The 1LUMO values for fluoroacenes are much larger than those for acenes and deutero-acenes. Possible superconducting transition temps. (Tcs) for the monoanions of deutero-acenes and fluoroacenes are larger than those for the monoanions of acenes. The transition temperature (Tc) value increases much more significantly by H-F substitution than by H-D substitution in acenes. The 1LUMO and Tc values significantly decrease with an increase in mol. size from the monoanions of 1f to 5f. The logarithmically averaged phonon frequencies (@ln) do not significantly change with an increase in mol. size in the monoanions of fluoroacenes. The larger displacements of C atoms in the vibronic active modes in fluoroacenes than those in deutero-acenes due to larger atomic mass of F than that of D, and the unchanged properties of the orbital patterns of the LUMO as a consequence of H-F and H-D substitution in acenes, are the main reasons why the 1LUMO value increases much more significantly by H-F substitution than by H-D

substitution, and the reason why the Tc value increases much more significantly by H-F substitution than by H-D substitution in acenes. The detailed properties of vibronic active modes and the electronic structures in the LUMO as well as the mol. wts. are closely related to the LLUMO, oln, and Tc values in the monoanions of fluoroacenes, deuteroacenes, and acenes.

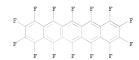
IT 646533-88-2

RL: PRP (Properties)

(carbon-carbon bond lengths in)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



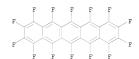
IT 645401-34-9

RL: PRP (Properties)

(electron-phonon interactions in)

RN 645401-34-9 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1-) (9CI) (CA INDEX NAME)



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http://www.cas.org/support/stngen/stndoc/properties.html

=> Uples

=> d 17 L7 HAS NO ANSWERS

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L7 STRUCTURE UPLOADED

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L7 STR F F F
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SAMPLE SCREEN SEARCH COMPLETED - 60 TO ITERATE
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100.0% PROCESSED 60 ITERATIONS 0 ANSWERS SEARCH TIME: 00.00.01

L8 0 SEA SSS SAM L7

=> s 17 full

FULL SEARCH INITIATED 09:46:59 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1124 TO ITERATE

100.0% PROCESSED 1124 ITERATIONS

SEARCH TIME: 00.00.01

11 SEA SSS FUL L7

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L10 0 L9 NOT L4

=> s 19 not pv > 2005 19 L9 3044168 PY > 2005 7 L9 NOT PY > 2005

=> s 19 not pv > 2004 19 L9

4248038 PY > 2004 4 L9 NOT PY > 2004

=> d 112 ibib abs hitstr 1-YOU HAVE REQUESTED DATA FROM 4 ANSWERS - CONTINUE? Y/(N):v

L12 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN 2004:577543 CAPLUS

ACCESSION NUMBER: DOCUMENT NUMBER: 141:260225

TITLE:

The essential role of H-F substitution in the electron-phonon interactions and electron transfer in the negatively charged acenes

AUTHOR(S):

Kato, Takashi; Yamabe, Tokio Fukui Institute for Fundamental Chemistry, Kyoto CORPORATE SOURCE: University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 121(5), 2356-2366

CODEN: JCPSA6; ISSN: 0021-9606 American Institute of Physics

DOCUMENT TYPE: Journal

PUBLISHER:

LANGUAGE: English

AR The single charge transfer through acenes, partially H-F substituted acenes, and fluoroacenes is discussed. The reorganization energies between the neutral mols. and the corresponding monoanions for partially H-F substituted acenes lie between those for acenes and fluoroacenes. The delocalization of the lowest unoccupied MOs (LUMO) by substituting hydrogen atoms by fluorine atoms with the highest electronegativity in every element is the main reason why the reorganization energy between the neutral mol. and the monoanion for partially H-F substituted acenes lies between those for acenes and fluoroacenes. This result implies that the neq, charged partially H-F substituted acenes would be better conductors with rapid electron transfer than the neg. charged fluoroacenes if we assume that the overlap of the LUMO between partially H-F substituted acenes is not significantly different from that between two neighboring fluoroacenes. The structures of the monoanions of acenes, fluoroacenes, and partially H-F substituted acenes are optimized under D2h geometry, and the Jahn-Teller effects in the monoanions of benzene and fluorobenzene are discussed. The vibration effect onto the charge transfer problem is also discussed. The C-C stretching modes around  $1500\ \mathrm{cm}{-1}$  are the main modes converting the neutral mols, to the monoanions in acenes, fluoroacenes, and partially H-F substituted acenes. It can be confirmed from the calculational results that the C-C stretching modes around 1500 cm-1 the most strongly couple to the LUMO in these mols. The main reason why the total electron-phonon coupling consts. (ILUMO) for the monoanions of acenes in which four outer hydrogen atoms are substituted by fluorine atoms are larger than those for the monoanions of acenes in which several inner hydrogen atoms are substituted by fluorine atoms is suggested. The relationships between the electron transfer and the electron-phonon interactions are discussed. The plot of the reorganization energies against the 1LUMO values is found to be nearly linear. In view of these results, the relationships between the normal and superconducting states are briefly discussed.

646533-88-2 690975-12-3 RL: PRP (Properties)

(essential role of H-F substitution in electron-phonon interactions and electron transfer in neg. charged acenes)

RN 646533-88-2 CAPLUS

ΙT

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT:

99 THERE ARE 99 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:480070 CAPLUS

DOCUMENT NUMBER: 141:197895

TITLE: Perfluoropentacene: High-Performance p-n Junctions and

Complementary Circuits with Pentacene

AUTHOR(S): Sakamoto, Youichi; Suzuki, Toshiyasu; Kobayashi,

Masafumi; Gao, Yuan; Fukai, Yasushi; Inoue, Youji;

Sato, Fumio; Tokito, Shizuo

CORPORATE SOURCE: Institute for Molecular Science, Okazaki, 444-8787, Japan

SOURCE: Journal of the American Chemical Society (2004),

126(26), 8138-8140

CODEN: JACSAT; ISSN: 0002-7863
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The authors report the synthesis and characterization of perfluoropentacene as an n-type semiconductor for organic field-effect transistors (OFETs). Perfluoropentacene is a planar and crystalline material that adopts a herringbone structure as observed for pentacene. OFETs with perfluoropentacene were constructed using top-contact geometry, and an electron mobility of 0.11 cm2 V-1 s-1 was observed Bipolar OFETs with perfluoropentacene and pentacene function at both neg. and pos. gate voltages. The improved p-n junctions are probably due to the similar

d-spacings of both acenes. Complementary inverter circuits were fabricated, and the transfer characteristics exhibit a sharp inversion of the output signal with a high-voltage gain.

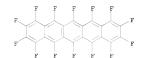
646533-88-2P, Perfluoropentacene RL: DEV (Device component use);

RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(perfluoropentacene: high-performance p-n junctions and complementary circuits with pentacene)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



L12 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: DOCUMENT NUMBER:

TITLE:

AUTHOR(S): CORPORATE SOURCE:

PUBLISHER:

SOURCE:

DOCUMENT TYPE: LANGUAGE:

2004:297458 CAPLUS 140:429323

> Inverse isotope effects and electron-phonon coupling in the positively charged deutero- and fluoroacenes Kato, Takashi; Yamabe, Tokio

Fukui Institute for Fundamental Chemistry, Kyoto University, Sakyo-ku, Kyoto, 606-8103, Japan

Journal of Chemical Physics (2004), 120(16), 7659-7672

CODEN: JCPSA6; ISSN: 0021-9606 American Institute of Physics

Journal English Electron-phonon interactions in the monocations of deutero- and

fluoroacenes are studied and compared with those in the monocations of acenes and those in the monoanions of fluoroacenes. Because of the significant phase pattern difference between the highest occupied MOs (HOMO) and the lowest unoccupied MOs (LUMO), the frequency modes lower than 500 cm-1 and the high-frequency modes around 1400 cm-1 couple more strongly to the LUMO than to the HOMO, while the frequency modes around 500 cm-1 and the frequency modes around 1600 cm-1 couple more strongly to the HOMO than to the LUMO in fluoroacenes with D2h geometry. The total electron-phonon coupling consts. for the monocations (1HOMO) are estimated and compared with those for the monoanions (1LUMO) in deutero- and fluoroacenes. The 1HOMO values are estimated to be 0.418, 0.399, 0.301, 0.255, and 0.222 eV for C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f), resp. The 1HOMO values are smaller than the 1LUMO values in small fluoroacenes. But the 1HOMO value decreases with an increase in mol. size less rapidly than the 1LUMO value in fluoroacenes, and the 1HOMO value of 0.074 eV is much larger than the 1LUMO value of 0.009 eV in polyfluoroacene. The logarithmically averaged phonon frequencies for the monocations (@ln, HOMO) are estimated to be larger than those for the monoanions (@ln,LUMO) in fluoroacenes. This is because the C-C stretching modes around 1600 cm-1 couple most strongly to the HOMO, and those around 1400 cm-1 couple the most strongly to the LUMO in fluoroacenes. The significant phase pattern difference between the HOMO and the LUMO is the main reason for the calculational results. The 1HOMO values increase much more significantly by H-F substitution than by H-D substitution in acenes. The possible inverse isotope effects in the electron-phonon interactions as a consequence of deuteration in the

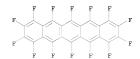
monocations of nanosized mols. are suggested. 690975-12-3

RL: PRP (Properties)

(inverse isotope effects and electron-phonon coupling in pos. charged deutero- and fluoroacenes)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT:

38 THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT ACCESSION NUMBER: 2003:898555 CAPLUS

DOCUMENT NUMBER: 140:118999

TITLE: Electron-phonon interactions in the monoanions of

fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Graduate School of Engineering, Department of Material

Chemistry, Kyoto University, Sakyo-ku, Kyoto,

606-8501, Japan

Journal of Chemical Physics (2003), 119(21),

11318-11328

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics DOCUMENT TYPE: Journal

LANGUAGE: English

Electron-phonon interactions in the monoanions of fluoroacenes such as C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f) are studied, and compared with those in the monoanions of acenes and deutero-acenes. The C-C stretching modes around 1500 cm-1 the most strongly couple to the lowest unoccupied MOs (LUMO) in fluoroacenes. The estimated total electron-phonon coupling consts. (ILUMO) are 0.475, 0.473, 0.350, 0.273, and 0.215 eV for 1f, 2f, 3f, 4f, and 5f, resp. The lLUMO values for fluoroacenes are much larger than those for acenes and deutero-acenes. Possible superconducting transition temps. (Tcs) for the monoanions of deutero-acenes and fluoroacenes are larger than those for the monoanions of acenes. The transition temperature (Tc) value increases much more significantly by H-F substitution than by H-D substitution in acenes. The 1LUMO and Tc values significantly decrease with an increase in mol. size from the monoanions of 1f to 5f. The logarithmically averaged phonon frequencies (mln) do not significantly change with an increase in mol. size in the monoanions of fluoroacenes. The larger displacements of C atoms in the vibronic active modes in fluoroacenes than those in deutero-acenes due to larger atomic mass of F than that of D, and the unchanged properties of the orbital patterns of the LUMO as a consequence of H-F and H-D substitution in acenes, are the main reasons why the 1LUMO value increases much more significantly by H-F substitution than by H-D substitution, and the reason why the Tc value increases much more significantly by H-F substitution than by H-D substitution in acenes. The detailed properties of vibronic active modes and the electronic structures in the LUMO as well as the mol. wts. are closely related to the 1LUMO, wln, and Tc values in the monoanions of fluoroacenes, deuteroacenes,

and acenes. 646533-88-2

RL: PRP (Properties)

(carbon-carbon bond lengths in)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)

645401-34-9

RL: PRP (Properties)

(electron-phonon interactions in)

RM 645401-34-9 CAPLUS

90

REFERENCE COUNT:

THERE ARE 90 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT